

TRIBOLOGICAL PROPERTIES OF SINTERED AUSTENITIC STAINLESS STEELS

Atrito

Desgaste

Lubrificação

Effects of boron, yttria, and other additives

V.04

Moqueca Tribológica

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Programa de Pós-Graduação em Engenharia e Ciência dos Materiais



Universidade de Caxias do Sul



Profa. María Cristina Moré Farias

- ❖ Graduação Eng. Mecânica – ISPJAE - Cuba
- ❖ Mestrado Eng. Mecânica – USP
- ❖ Doutorado Eng. Mecânica – USP
- ❖ Pós-Doutorado Eng. Mecânica – USP
- ❖ **Docente – PPGMAT/UCS (2010 - atual)**

Principais Pesquisas

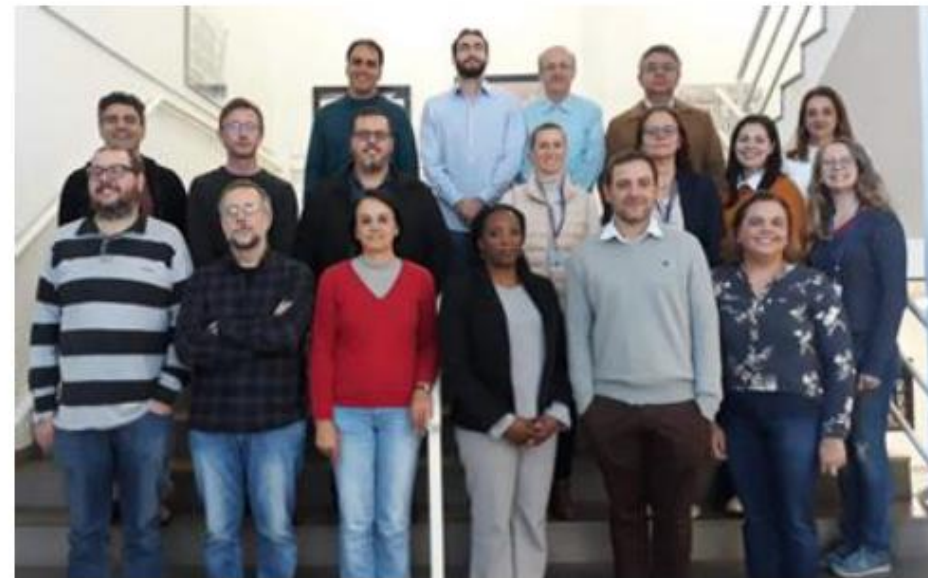
- ❖ Desenvolvimento e caracterização tribológica materiais sinterizados (metais, cerâmicas) com diferentes aditivos (ativadores, reforços, lubrificantes sólidos)
- ❖ Tribologia de materiais de fricção utilizados em freios automotivos
 - ❖ Determinação de propriedades mecânicas de superfícies empregando indentação instrumentada
- ❖ Desenvolvimento de pavimentos cerâmicos base argila com adição de resíduos de rochas

PPGMAT - UCS

Programa de Pós-Graduação em Engenharia e Ciência dos Materiais
Universidade de Caxias do Sul

Docentes 2019

Alexandre Fassini Michels
Carlos Alejandro Figueroa
César Aguzzoli
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Janete Eunice Zorzi
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Marcelo Giovanela
Márcio Ronaldo Farias Soares
Maria Cristina Moré Farias
Mariana Roesch Ely
Otávio Bianchi
Robinson Carlos Dudley Cruz
Sidnei Moura e Silva
Thiago Barcellos da Silva



Professores PPGMAT 2019

<https://www.ucs.br/site/pos-graduacao/formacao-stricto-sensu/materiais/>

PPGMAT - UCS

NÚMEROS

- Corpo discente 2019: 17 alunos de mestrado e 26 de doutorado, além de estudantes de graduação realizando trabalhos de iniciação científica.

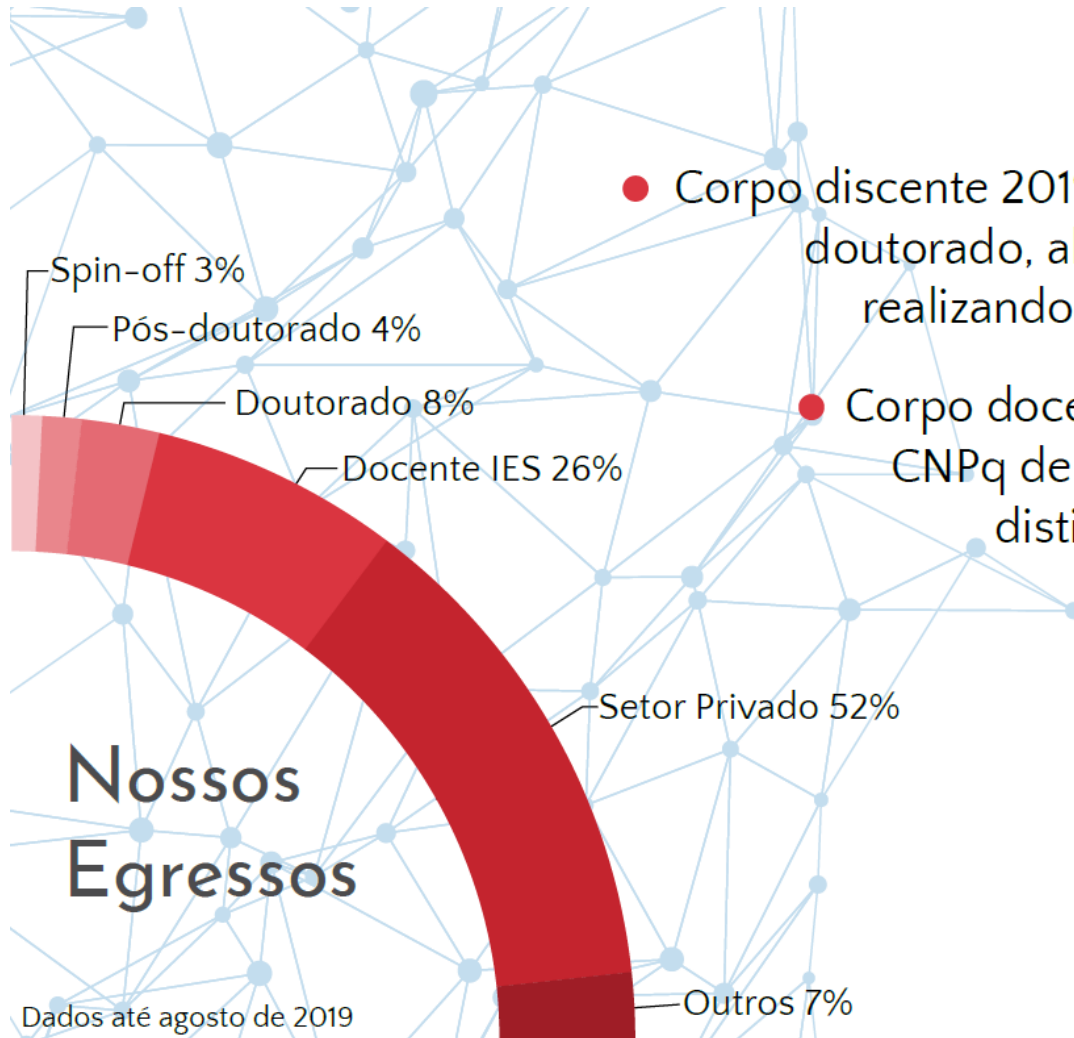
- Corpo docente permanente: 75% têm bolsa CNPq de Produtividade em Pesquisa, que distingue os pesquisadores com alta produtividade.

- 129 mestres formados.

- 24 doutores formados.

- 45 patentes.

- 570 artigos.



<https://www.ucs.br/site/pos-graduacao/formacao-stricto-sensu/materiais/>

PPGMAT - UCS

LINHA DO TEMPO



2003 • Prof. Israel J. R. Baumvol é convidado a criar um Programa de Pós-Graduação em Materiais

2004 • Início das atividades do PPGMAT com 85 inscritos para 15 vagas

2005 • Primeira defesa de Mestrado, pela tecnóloga em Polímeros Maira Finkler

2006 • Convênio UCS-SIMECS para complementar recursos públicos para laboratório do PPGMAT

- Inauguração do Laboratório de Engenharia de Superfícies e Tratamentos Térmicos (LESTT)
- Início do Doutorado Interinstitucional PGCIMAT/UFRGS - PPGMAT/UCS
- Inauguração do Laboratório de Caracterização de Materiais I

2008 • O PPGMAT compõe a recém criada "Área de Materiais" da CAPES
• Inauguração do Núcleo de Pesquisas em Geoquímica (NupGeo)

- Criação do Instituto Nacional de Engenharia de Superfícies (INES)
- Convênio UCS-CIC para complementar recursos públicos ao laboratório do INES
- Início das atividades do Laboratório de Pesquisa em Química dos Materiais (LPQM)

<https://www.ucs.br/site/pos-graduacao/formacao-stricto-sensu/materiais/>

PPGMAT - UCS



- O PPGMAT obtém nota 5 na avaliação trienal da CAPES
- Inauguração do Laboratório de Caracterização de Materiais para Mineração

2013

- Inauguração do Laboratório de Reologia
- O IMC integra a rede de laboratórios associados ao INMETRO para Inovação
- Comemoração dos 10 anos do PPGMAT, com palestra do professor Deniol Tanaka

2014

- Inauguração do Laboratório Central de Microscopia Prof. Israel Baumvol
- Segunda Edição do Brafitec 2015-2018

2015

- Primeira defesa de Doutorado do PPGMAT de Ana Cláudia Rangel Faria
- Primeira cerimônia de Titulação da Pós-Graduação

2016

- Assinatura de Convênio de Consolidação de Internacionalização com Lodz University of Technology

2017

- Participação no Fórum BRAFITEC com visita a instituições parceiras.

2018

- Assinatura de convênios internacionais - Programa Erasmus+ e Athlone Institute of Technology

2019

<https://www.ucs.br/site/pos-graduacao/formacao-stricto-sensu/materiais/>

Sintered austenitic stainless steels

- Stainless steels have been successfully fabricated through different powder metallurgy (P/M) routes
- Austenitic (ASS) and ferritic (FSS) stainless steels are the most widely produced by P/M
- ASS exhibit a good combination of corrosion and oxidation resistance, associated with good mechanical properties
- Interest in P/M SS for general use (biomedical, dental, chemical, nuclear, automotive, aerospace) has increased

Sintered austenitic stainless steels

- P/M ASS present lower mechanical resistance than the wrought or cast steels, due to their intrinsic porosity
- Applications of ASS are also limited by their relative softness and susceptibility to wear (adhesive, abrasive, fatigue) and wear-corrosion
- Three routes have been implemented to improve density and reach a good combination of mechanical, wear and corrosion properties of P/M SS
 - i. surface modification of the sintered body by plasma-assisted surface treatments
 - ii. modification of parameters in compaction and sintering steps
 - iii. addition of certain elements (sintering enhancers or activators and reinforcements)

Activated sintering

- **Activated sintering refers to any special process which results in an increased sintering rate or densification rate, i.e.,**
 - promotes lowering sintering temperature; shorten sintering time or improve sintered properties
- **Sintering enhancement approaches**
 - Solid state activated sintering
 - Liquid phase sintering
- **Sintering activators**
 - small particles, frequently used in low concentrations
 - promote effective changes in interfacial energy, grain boundary mobility, reduction of void fraction, diffusion rates, and phase stability

R. M. German & B. H. Rabin (1985) Powder Metallurgy, 28:1, 7-12.

Possible routes to alter sintering rate

- Change process conditions (particle size or temperature)
- Change defects configuration by pretreating powders (alloying or deformation)
- Application of external force (Ex. HP, HIP)
- Promote the formation of second phases that act as preferential diffusion paths
 - Solid state activated sintering
 - Liquid phase sintering

R. M. German & B. H. Rabin (1985) Powder Metallurgy, 28:1, 7-12.

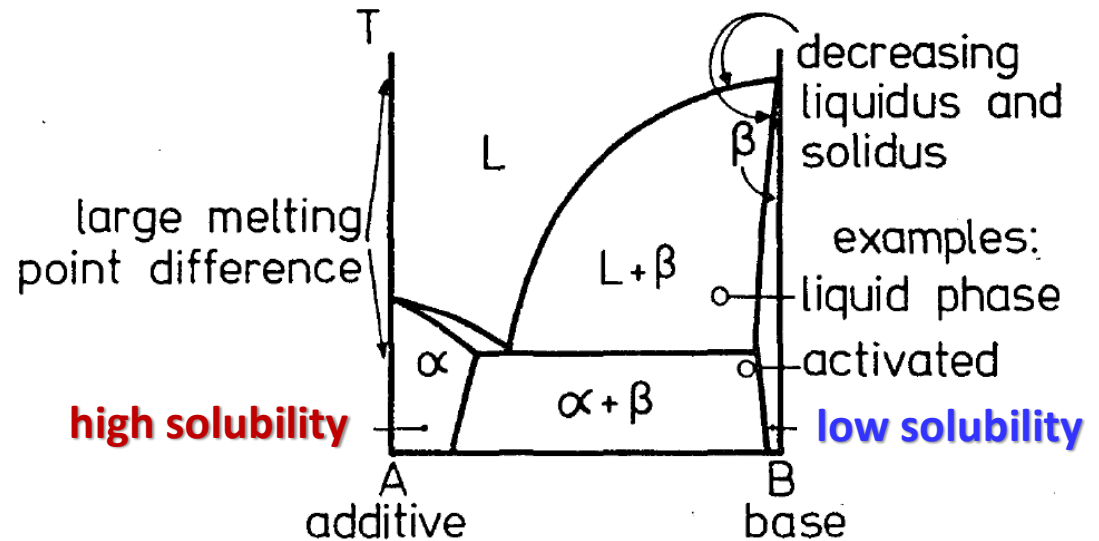
Liquid phase sintering

- **In the case of liquid-phase sintering, densification is achieved through the formation of a system with high wettability between the matrix and the liquid phase**
 - improves the mass transfer rate
 - Increase the sintering rate by decreasing sintering temperature or reducing sintering time
- **Liquid phase can be obtained by**
 - **addition of low melting temperature elements**
 - Cu, P, Si, Cu-10Sn, Tin, Babbitt
 - **dissociation of a mixture containing the base material and additive powders in a new phase with eutectic composition**
 - B, Cr₂B, FeB, Fe₂B

R. M. German & K. A. D'Angelo (1984), *International Metals Reviews*, 29:1, 249-272.

Criteria for selecting sintering additive

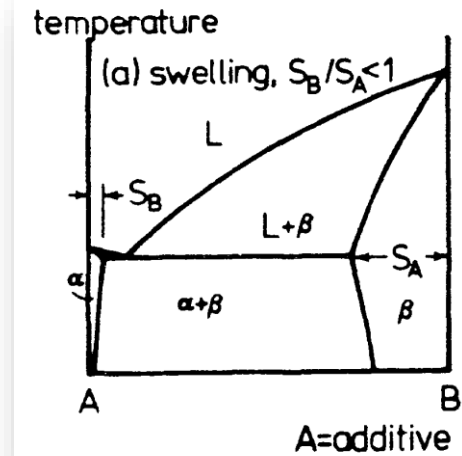
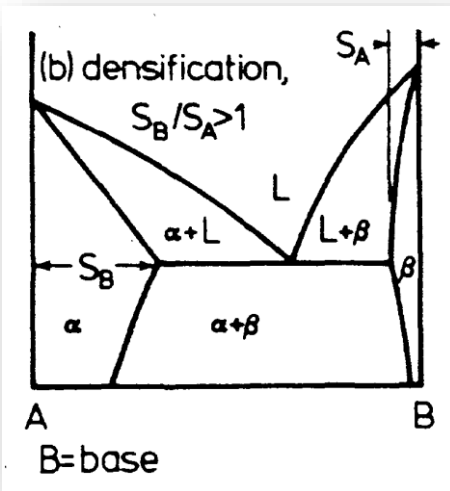
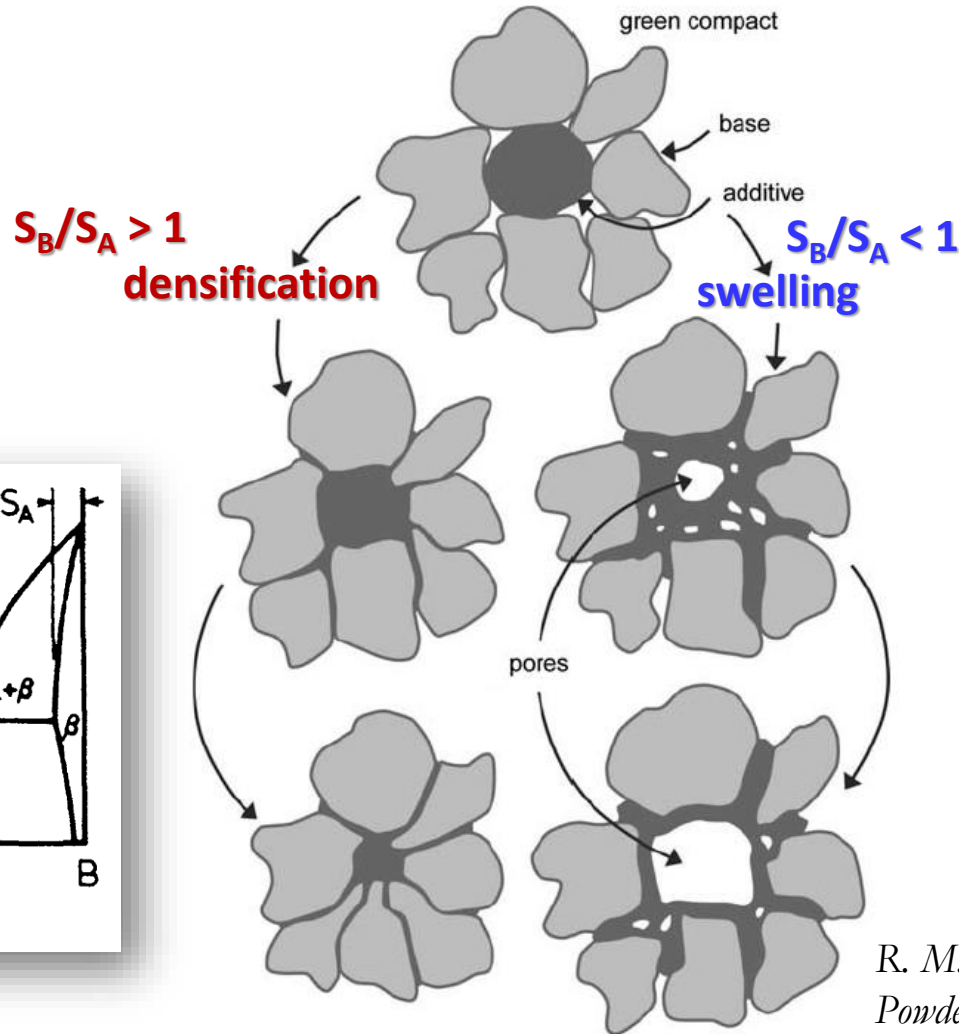
- **Solubility: $S_B/S_A > 1$**
 - Favorable effect in diffusion rate
 - diffusive flux in additive layer
 - favorable change in bonding free energy
 - good wetting and adhesion of additive to base material



Idealized phase diagram showing characteristics most favorable for enhanced sintering

R. M. German & K. A. D'Angelo (1984), *International Metals Reviews*, 29:1, 249-272.

Criteria for selecting sintering additive



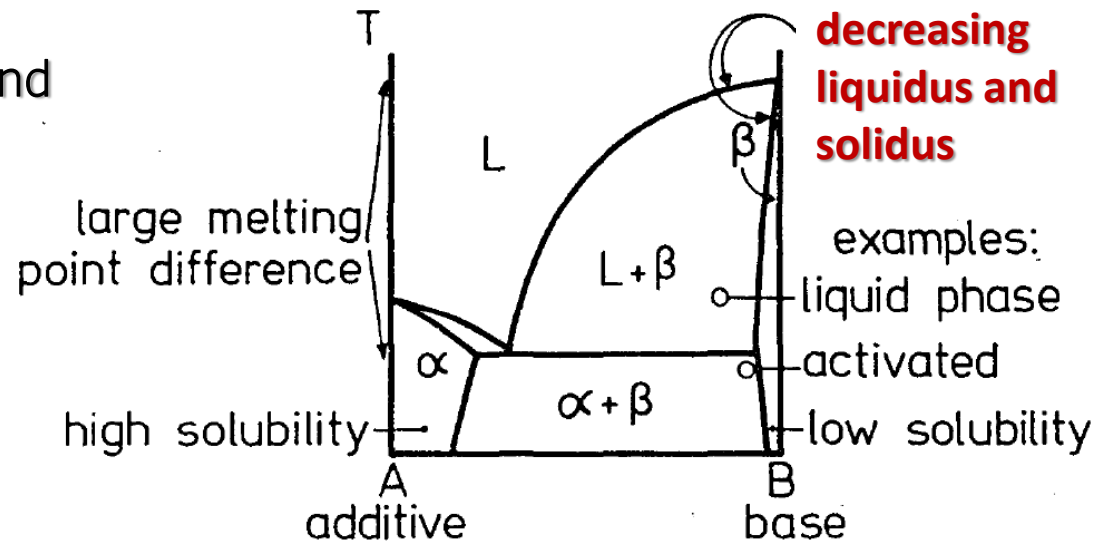
R. M. German & B. H. Rabin (1985)
Powder Metallurgy, 28:1, 7-12.

Criteria for selecting sintering additive

- **Segregation: $T_{mB}/T_{LP} > 1$**

- Segregation of an equilibrium second phase at interparticle site

- Decreasing liquidus and solidus as A is increased

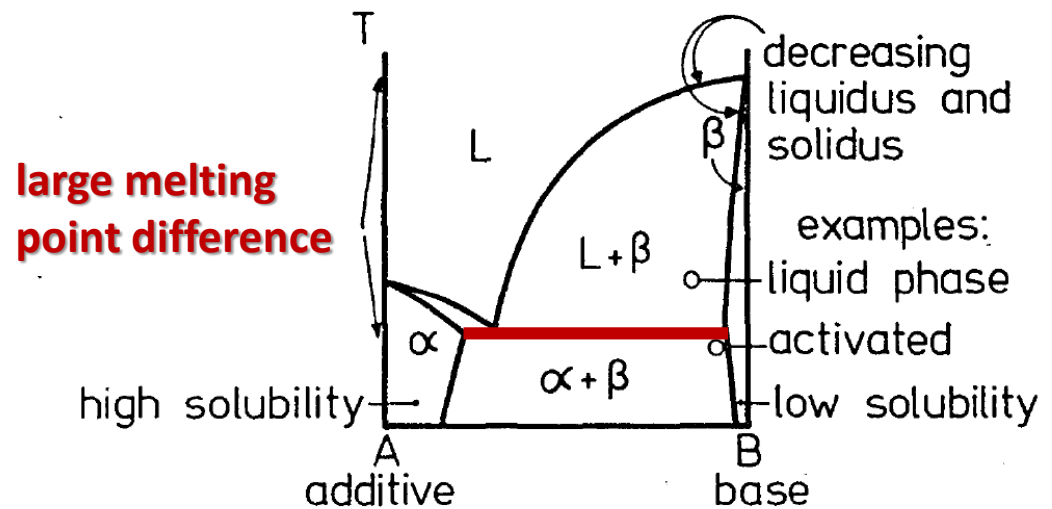


R. M. German & B. H. Rabin (1985) *Powder Metallurgy*, 28:1, 7-12.

Criteria for selecting sintering additive

● Diffusion: $D_E/D_B > 1$

- D_E : diffusivity of B in A layer
- D_B : selfdiffusivity of B
- additive flows to the interparticle boundary
- rapid diffusion along the sinter bond
- low liquid temperature for A
 - low activation energy
 - high diffusivity



R. M. German & B. H. Rabin (1985) *Powder Metallurgy*, 28:1, 7-12.

Sintering additives for ferrous powders

- Sintering activators
 - C, B, P,
 - Cu, Sn, S, Ni, Mn, Co, Ti
- Other additives
 - Al_2O_3 , Y_2O_3
 - B_2Cr , Cr_2Al , TiCr_2 , TiAl
 - VC, SiC, TiC
 - TiB_2

R. M. German & K. A. D'Angelo (1984), *International Metals Reviews*, 29:1, 249-272.

Oke, S.R., Ige, O.O., Falodun, O.E. et al. (2019) *Int J Adv Manuf Technol*, 102, 3271–3290.

Other sintering additives

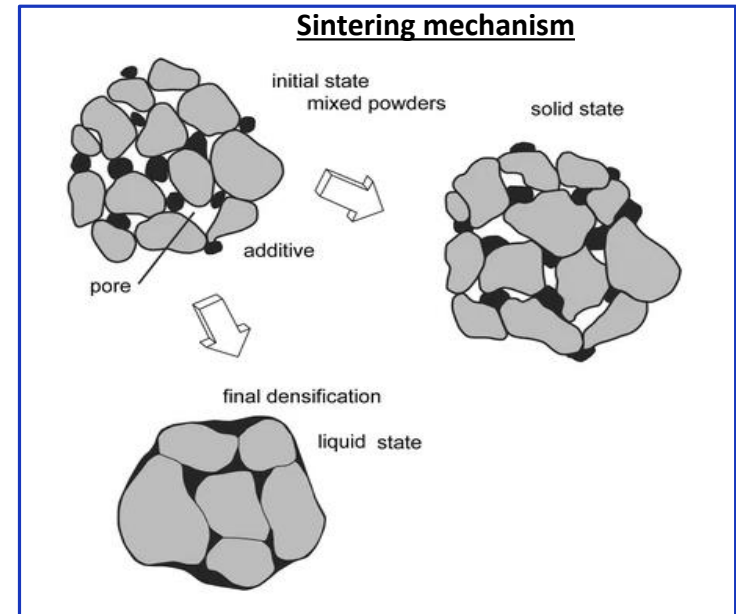
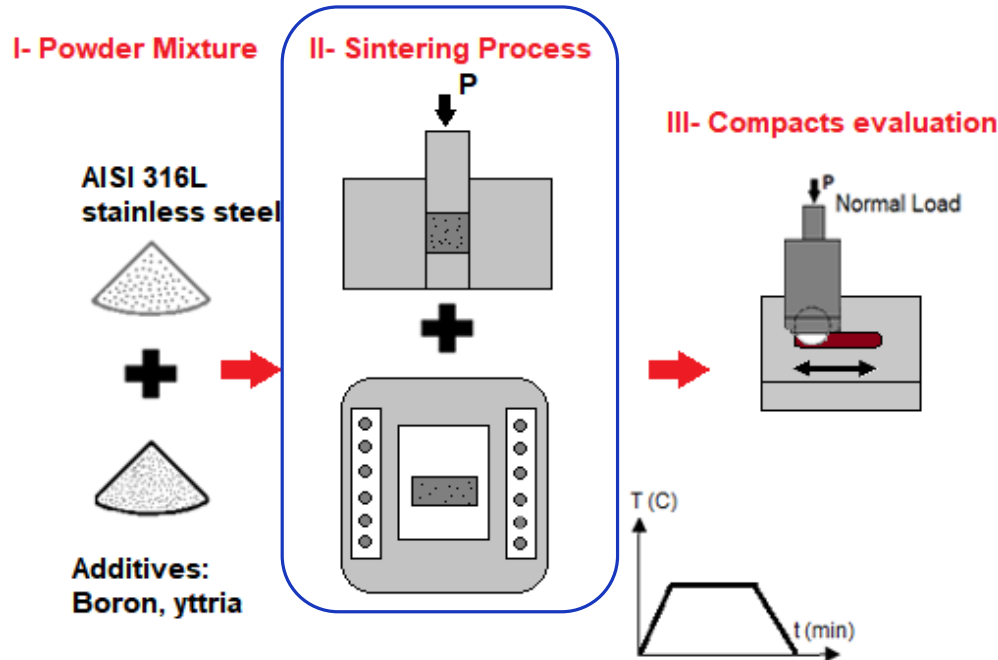
- In combination with solid state or liquid phase sintering activators, other elements have been added to
 - enhance mechanical properties
 - improve corrosion and high-temperature oxidation resistance
 - reduce friction and wear
- Low matrix-additive interaction makes necessary the use sintering activators
 - Solid lubricants (h-BN, MoS₂)
 - Reinforcement
 - Oxide ceramics (Al₂O₃, Y₂O₃) → mechanical properties, wear and corrosion resistance
 - Intermetallics (Cr₂Al, TiCr₂, TiAl) → mechanical, corrosion and wear properties
 - Carbides (VC, SiC, TiC) → mechanical properties, wear resistance
 - Borides (TiB₂) → mechanical and tribological properties
- The improvement of mechanical, corrosion and tribological properties of the composites depends on the amount, size, shape and distribution of the dispersed second phase particles, and P/M parameters

Oke, S.R., Ige, O.O., Falodun, O.E. et al. (2019) Int J Adv Manuf Technol, 102, 3271–3290.

Tribology properties of sintered austenitic stainless steels

- The literature on the tribological properties of P/M austenitic stainless steels and their composites is scarce
- There exist some researches on dry sliding behavior of sintered austenitic stainless steels added with
 - Metals or metallic alloys: B, Cu-Sn
 - Borides: TiB_2
 - Nitrides: BN
 - Oxides: Al_2O_3 , Y_2O_3 , YAG
 - Carbides: SiC, VC
 - Intermetallics: $TiCr_2$, Cr_2Al , Ni_3Al , Fe_3Al

Metal matrix composites (MMC)



Ref: German et al Review: liquid phase sintering. *J Mater Sci* (2009) 44:1–39

Powder metallurgy

- Effect of sintering parameters as temperature on the sintered phase transformation
- Importance of the particles: size, shape and volume fraction
- Effect of mixture of different materials

Secondary particle additives

Sintering mechanism

- Liquid phase sintering
- Solid phase sintering

Final Properties

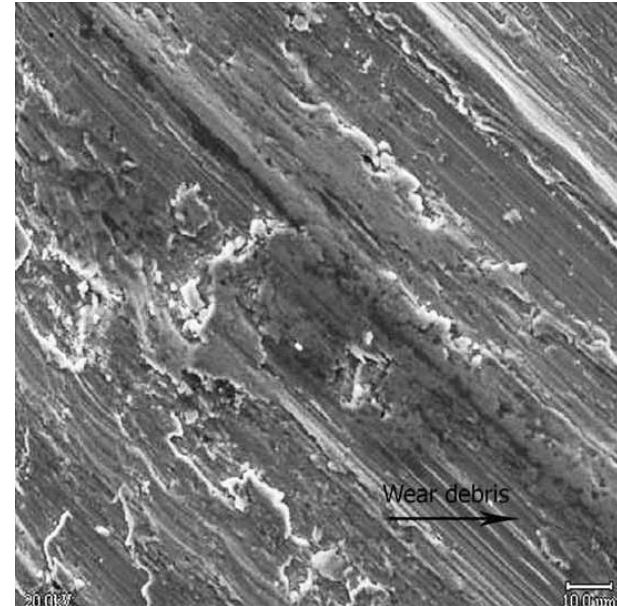
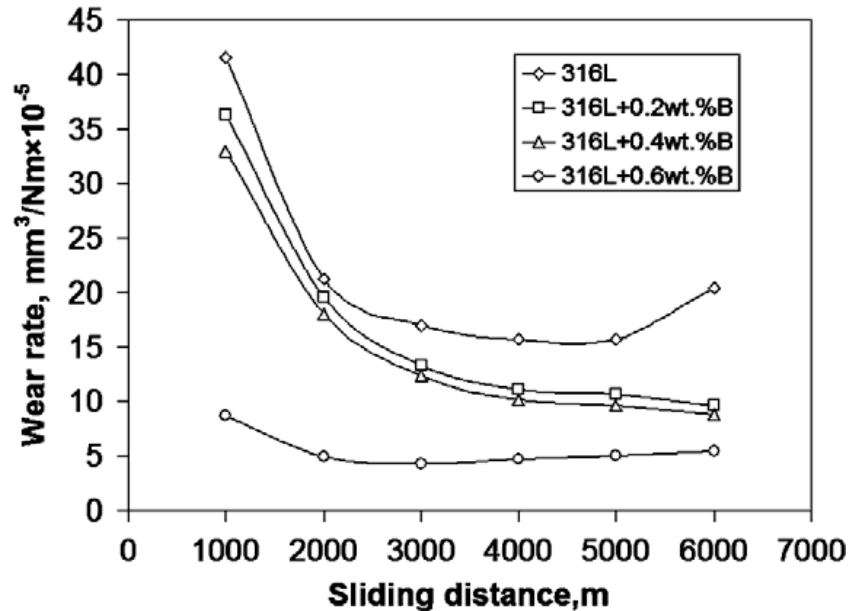
- Higher hardness and density
- Oxidation resistance
- Wear resistance

Sintered Austenitic Stainless Steel

Precedents

- D. Uzunsoy, *Investigation of dry sliding wear properties of boron doped powder metallurgy 316L stainless steel*, *Mater. Des.* 31 (8) (2010) 3896–3900.
- M. Vardavoulias, M. Jeandin, F. Velasco, J.M. Torralba, *Dry sliding wear mechanism for P/M austenitic stainless steels and their composites containing Al_2O_3 and Y_2O_3 particles*, *Tribol. Int.* 29 (6) (1996) 499–506.
- A. Bautista, F. Velasco, J. Abenojar, *Oxidation resistance of sintered stainless steels: effect of yttria additions*, *Corros. Sci.* 45 (2003) 1343–1354.

Dry sliding wear of boron doped P/M ASS (Uzunsoy, 2010)



- ❑ Boron additions decrease plastic deformation and wear rate in sliding contact
- ❑ Hardness and porosity level have a significant effect on the wear behavior of P/M ASS

Sliding wear mechanism for P/M ASS and their composites (Vardavoulias et al., 1996)

Table 3 Pin-on-disc wear testing results (alumina counterbody)

Specimen	Friction coefficient	Disc specific wear rate ($\times 10^{-13} \text{ m}^2 \text{ N}^{-1}$)
304L	0.61	7
304L+B ₂ Cr+Y ₂ O ₃	0.62–0.65	5.7
304L+B ₂ Cr+Al ₂ O ₃	0.8	2.1
304L+BN+Y ₂ O ₃	0.65–0.68	9.4
304L+BN+Al ₂ O ₃	0.6–0.7	6.1
316L	0.58	6.1
316L+B ₂ Cr+Y ₂ O	0.6–0.65	3.5
316L+B ₂ Cr+Al ₂ O ₃	0.6–0.65	3.4
316L+BN+Y ₂ O ₃	0.62–0.65	4.7
316L+BN+Al ₂ O ₃	0.62–0.65	4.8

- Ceramic particles (Al₂O₃ and Y₂O₃) and sintering activators (B₂Cr, BN) improved wear resistance
- Ceramic particles limited plastic deformation while sintering activators decreased porosity
- Friction coefficient did not vary substantially (0.6 and 0.7)
- “Friction-induced martensite” (debris)

Oxidation resistance of sintered stainless steels (Bautista et al., 2003)

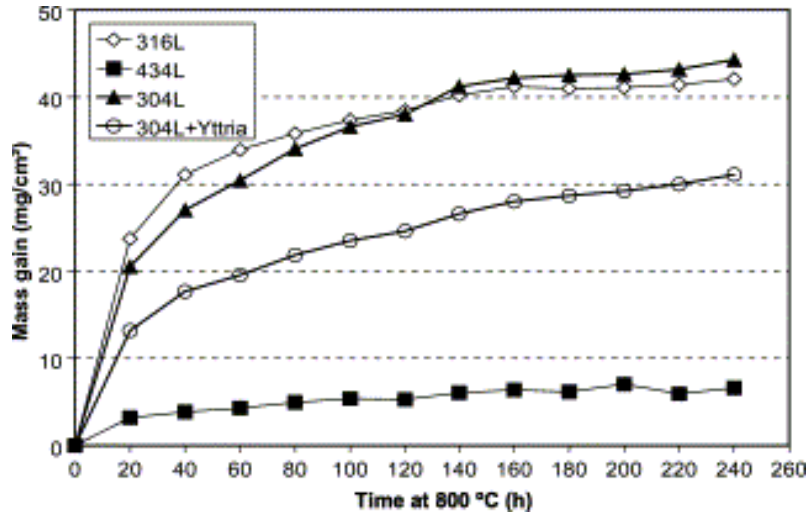


Table 2. Parabolic kinetic constants

Sintered stainless steels	k ($\text{g}^2 \text{cm}^{-4} \text{s}^{-1}$)	
	600 °C	800 °C
316L	–	1.4E-10
304L	2.4E-11	8.1E-11
304L + Yttria	2.4E-11	3.6E-10
434L	4.0E-12	2.5E-12

Sintered Austenitic Stainless Steel – recent results

- ❖ Friction and wear
- ❖ High-temperature oxidation

Effect of boron and yttria additions

Team

Profa. M. Cristina Moré Farias

Postgraduate students

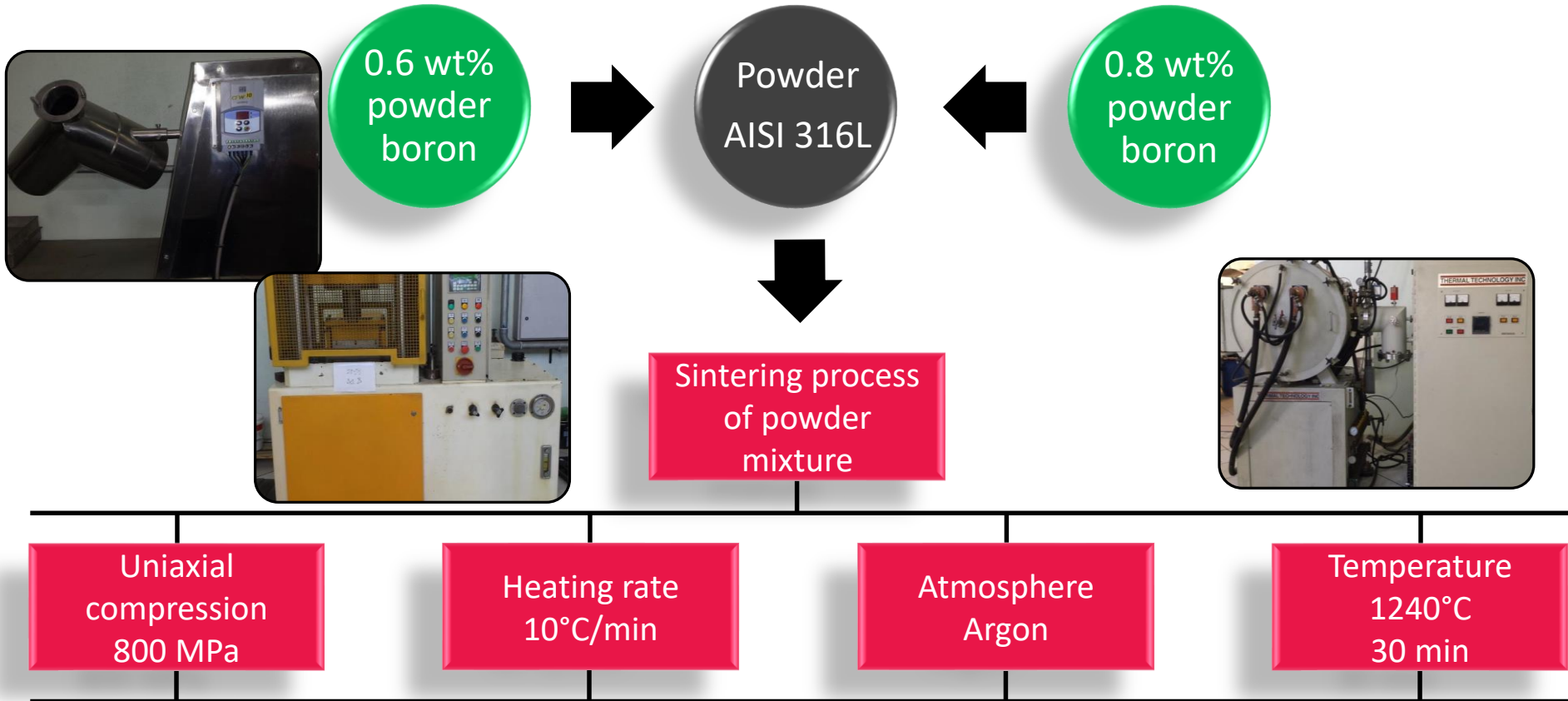
Collaborators professors

- Serafini, F.L., Peruzzo, M., Krindges, I., Ordoñez, M.F.C., Rodrigues, D., Souza, R.M., Farias, M.C.M. (2019) *Materials Characterization*, 152, pp. 253-264.
- Peruzzo, M., Serafini, F.L., Ordoñez, M.F.C., Souza, R.M., Farias, M.C.M. (2019) *Wear*, 422-423, pp. 108-118.
- Peruzzo, M., Beux, T.D., Ordoñez, M.F.C., Souza, R.M., Farias, M.C.M. (2017) *Corrosion Science*, 129, pp. 26-37.
- Serafini, F. L. ; Peruzzo, M. ; Beux, T. D. ; Ordoñez, M.F.C. ; Dotta, A. L. B. ; Souza, R.M. ; Farias, M.C.M. In: 6th World Tribology Congress, WTC2017, Beijing. *6th World Tribology Congress*, WTC2017, 2017.

Further studies for P/M ASSs and their composites

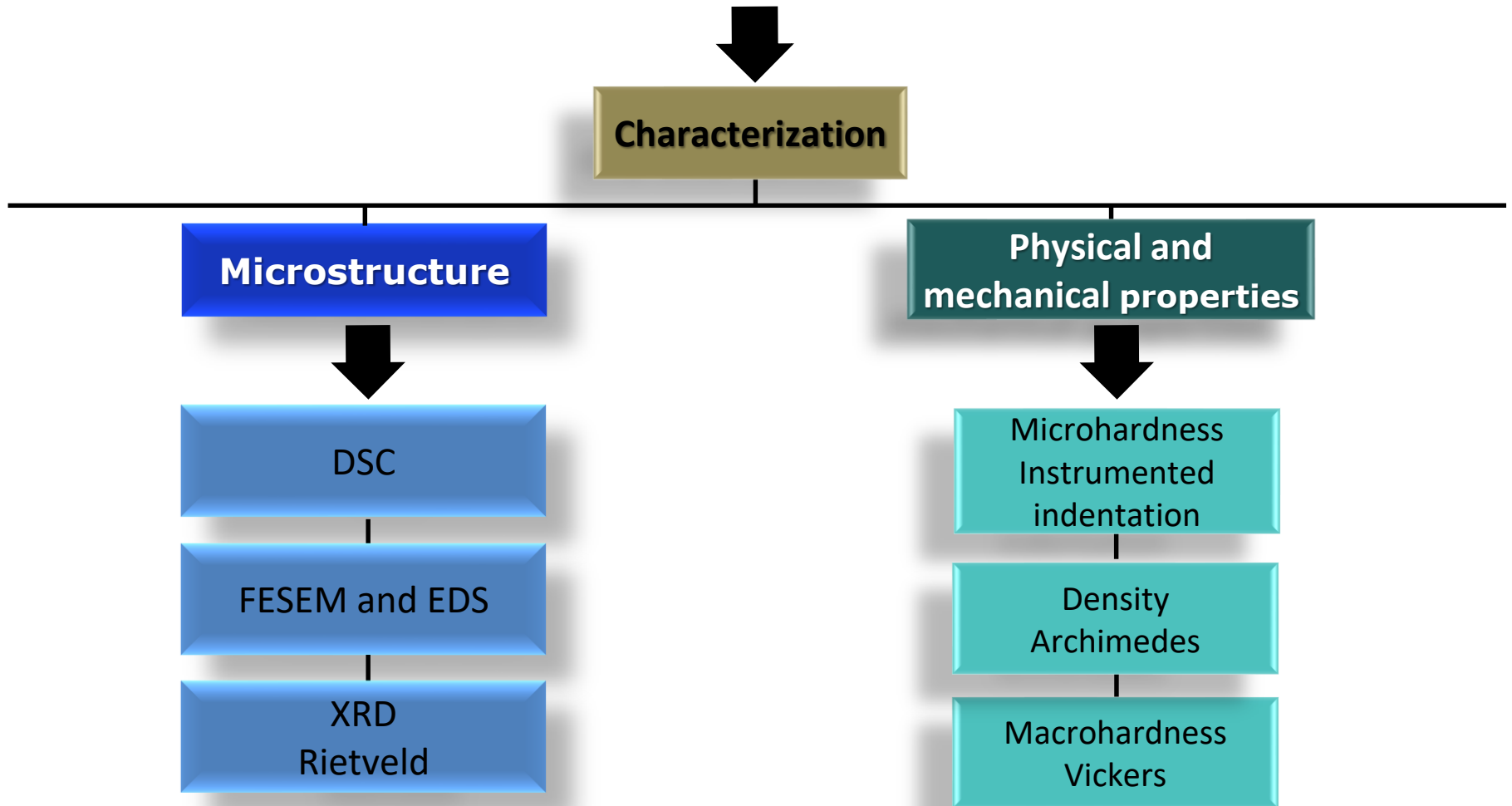
- Friction and wear at high-temperature
- Tribocorrosion behavior

P/M processing of ASS 316L



BRATS - Sintered Filters Special Metallic Powders (Cajamar, Sao Paulo, Brazil)

Initial characterization of the sintered ASS 316L samples



Tribological characterization of the sintered ASS 316L samples

Characterization

Tribological Properties

Reciprocating Sliding

3 samples
Alumina ball

Load 8 N
3 Hz

Time: 7200 s
Length: 4 mm

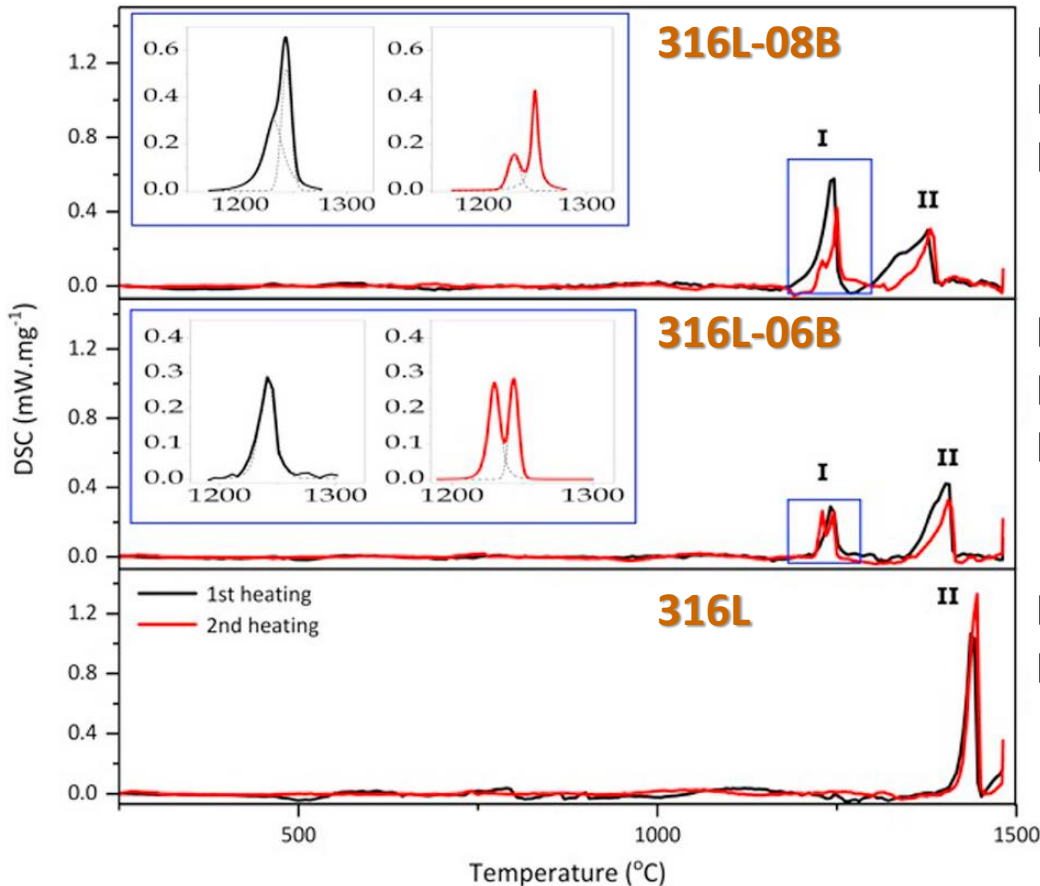
Friction
Coefficient

Wear rate
Wear mechanisms

UMT TriboLab
Bruker



Thermal analysis



I – Eutectic reaction 1219 °C – 1254 °C
II - Melting onset 1335 °C
II – Complete melting at 1385 °C

I – Eutectic reaction 1223 °C - 1251 °C
II – Melting onset at 1351 °C
II – Complete melting at 1412 °C

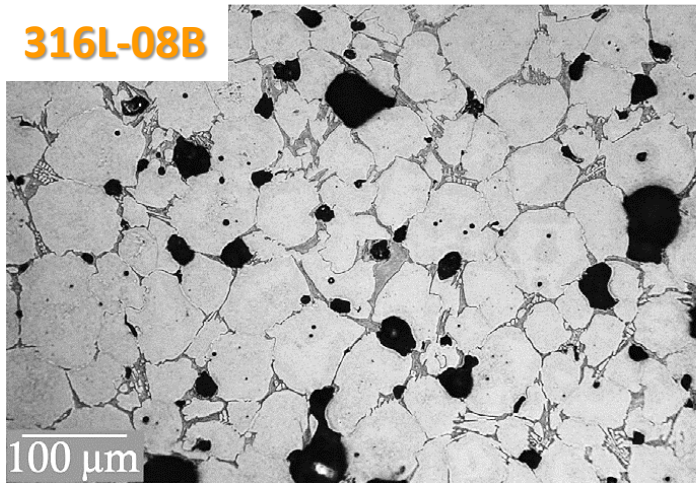
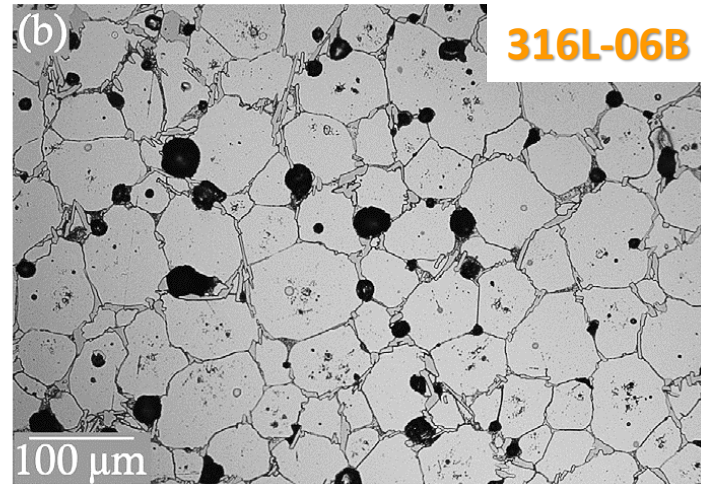
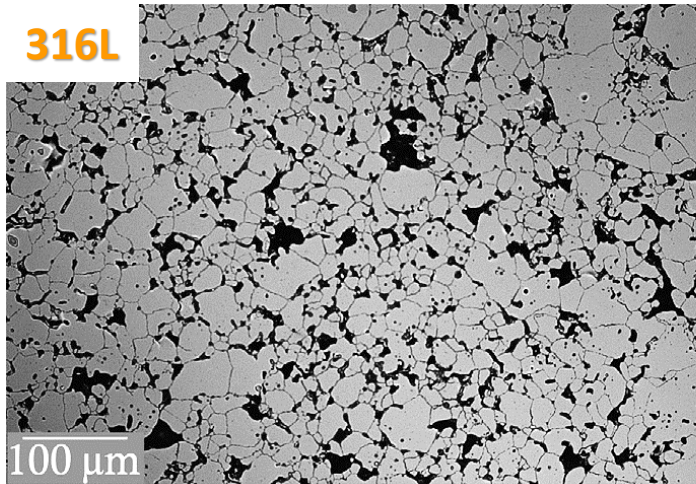
II – Melting onset at 1419 °C
II - Complete melting at 1451 °C

Boron addition provides

- ❖ decrease in liquid phase temperature
- ❖ greater amount of liquid phase

Serafini, F.L. et al. (2019)
Materials Characterization, 152, pp. 253-264.

Microstructure

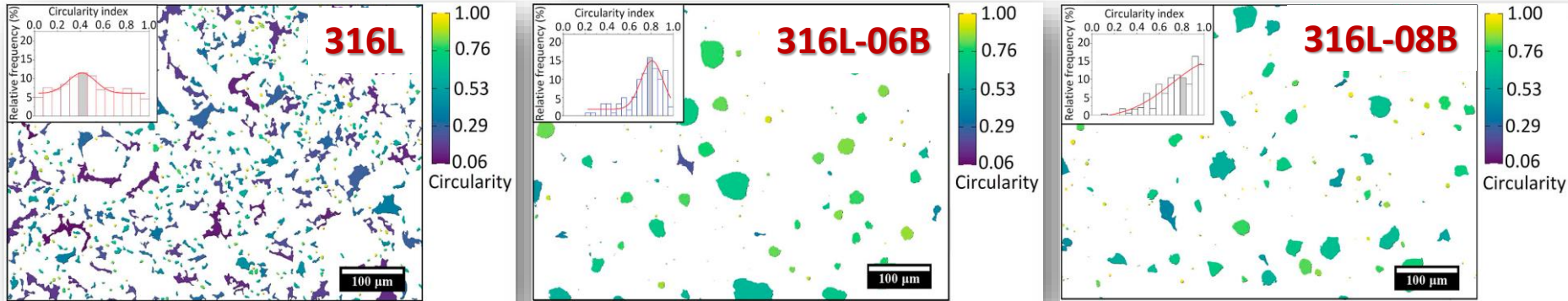


Microstructure of the B-free and B-containing sintered samples

- ❖ Porosity
- ❖ Austenitic grains
- ❖ Phases at the grain boundaries

Peruzzo et al. (2019) Wear, 422-423, pp. 108-118.

Porosity analysis

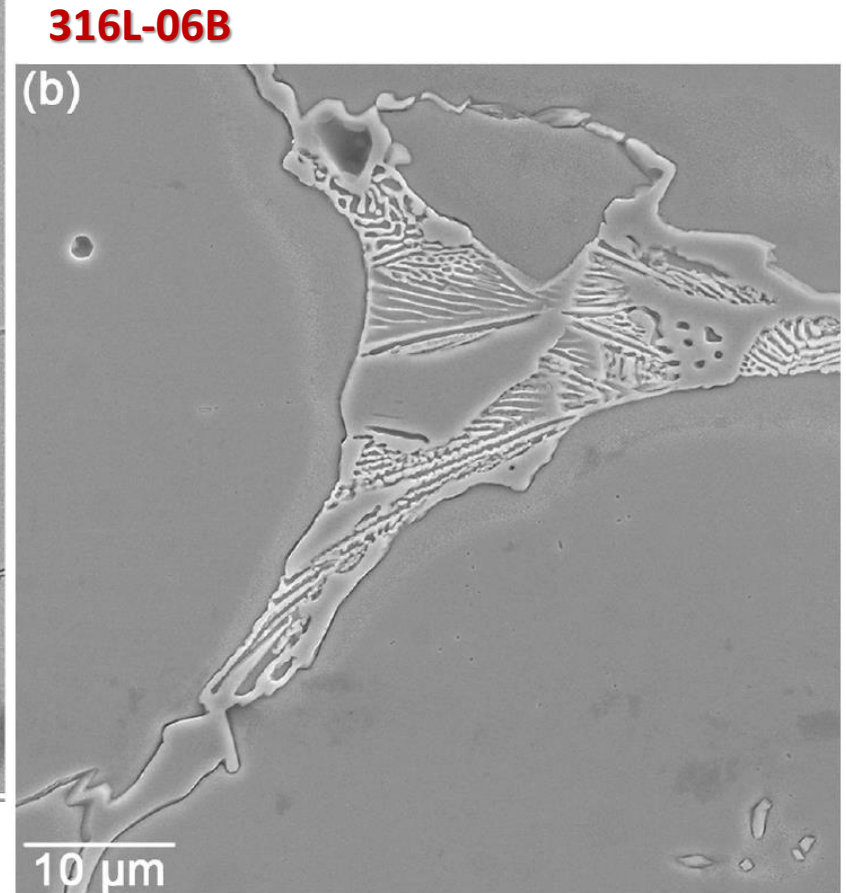
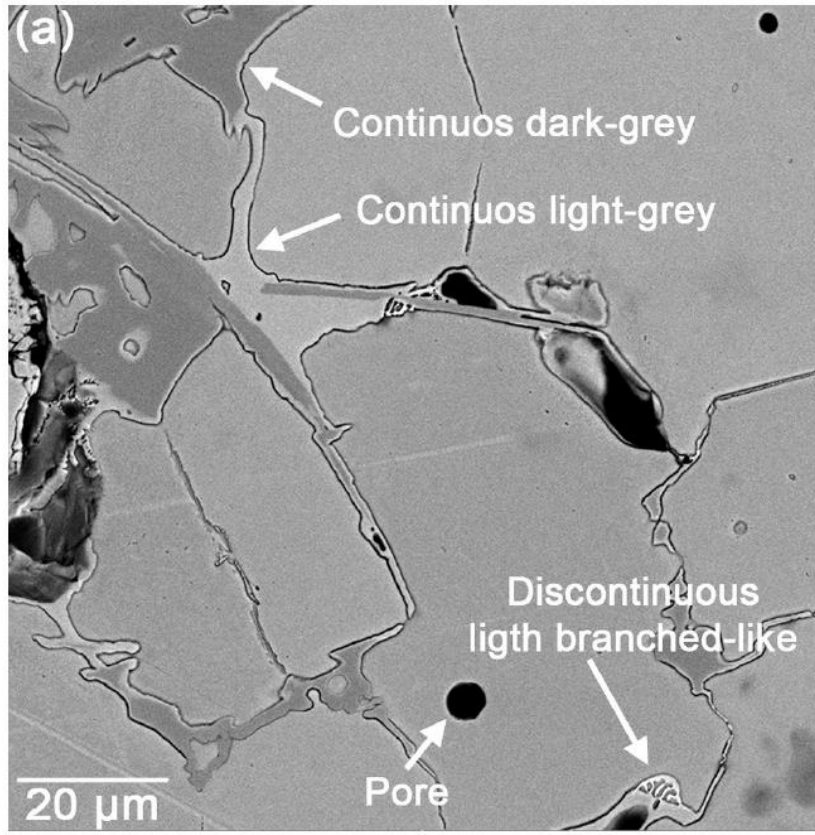


Sample	Porosity (%)	Pore size, D_{90} (μm)	Circularity index, C_i
316L	12.89	14.51	0.4
316L-06B	5.66	20.51	0.8
316-08B	7.7	26.17	0.8

- ❖ Irregular and interconnected pores were formed for the boron-free sample
- ❖ Nearly circular and isolated pores were developed in the boron-containing samples

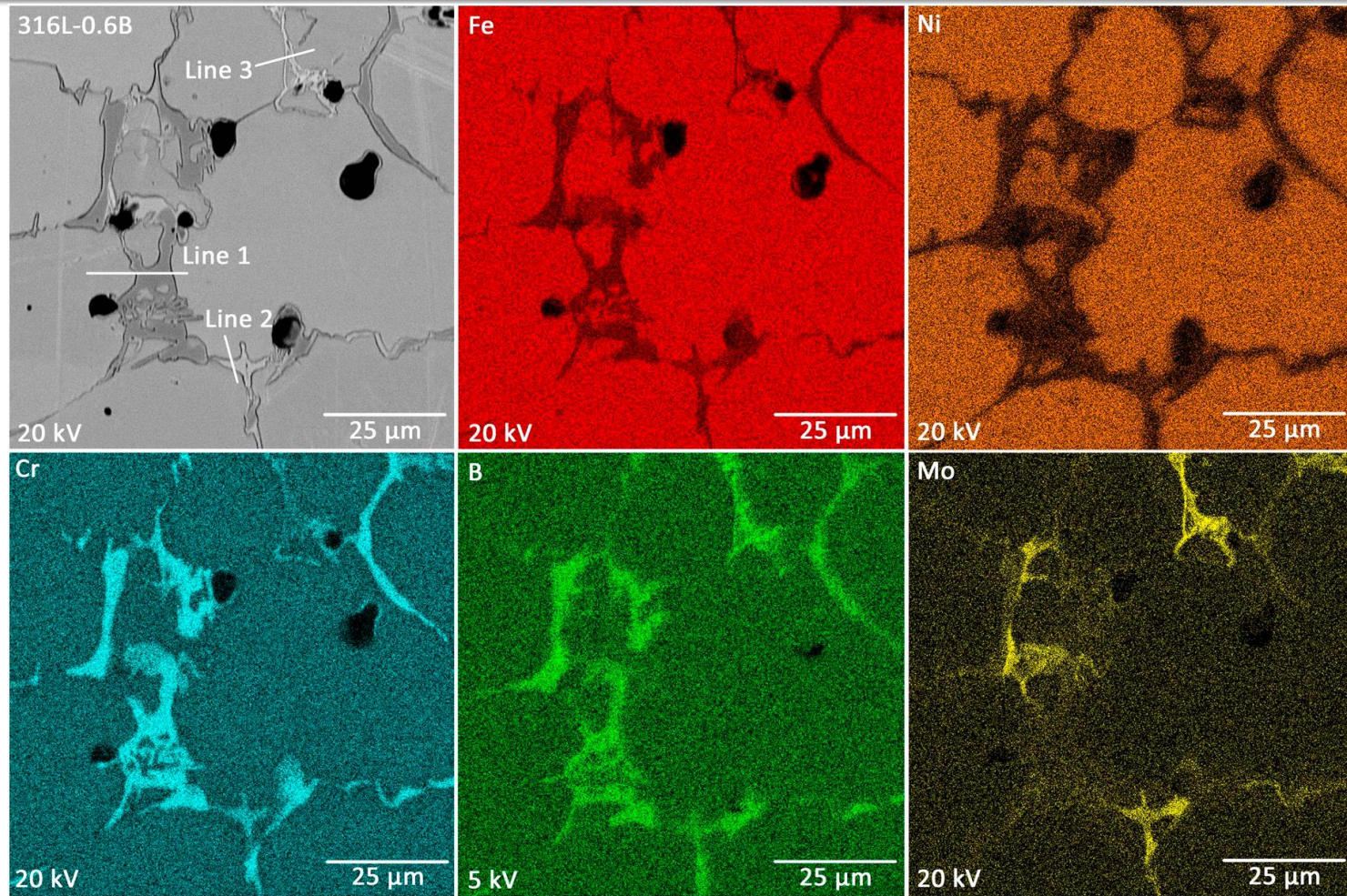
Serafini, F.L. et al. (2019) *Materials Characterization*, 152, pp. 253-264.

Microstructure



Serafini, F.L. et al. (2019) *Materials Characterization*, 152, pp. 253-264.

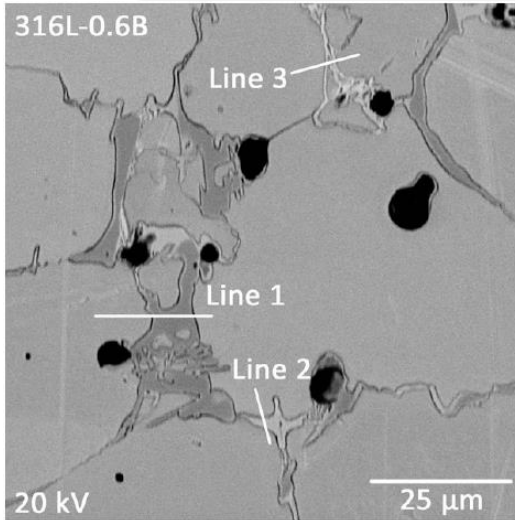
Microstructure – EDS mapping



Serafini, F.L. et al. (2019) *Materials Characterization*, 152, pp. 253-264.

Microstructure – EDS line

Serafini, F.L. et al. (2019) *Materials Characterization*, 152, pp. 253-264.



Line 1 – Continuous dark-grey phase, Cr-rich boride

Orthorhombic, $\text{Fe}_{1.1}\text{Cr}_{0.9}\text{B}_{0.9}$

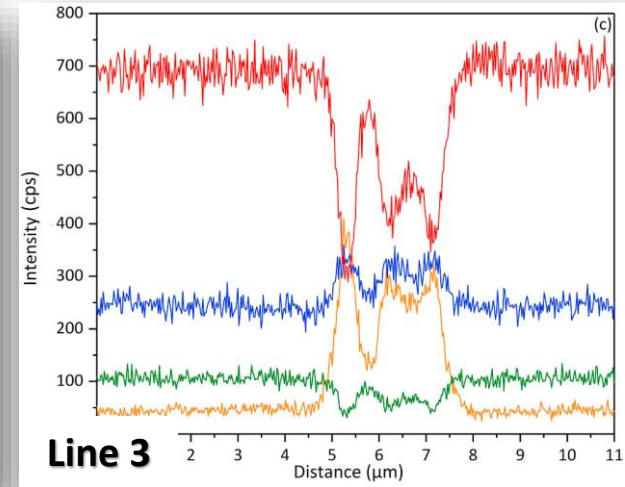
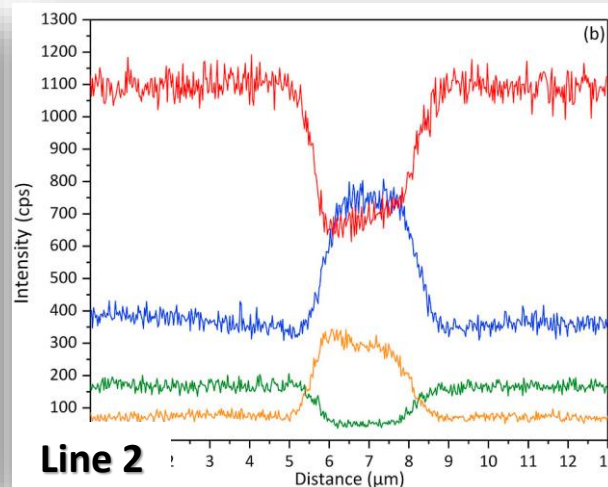
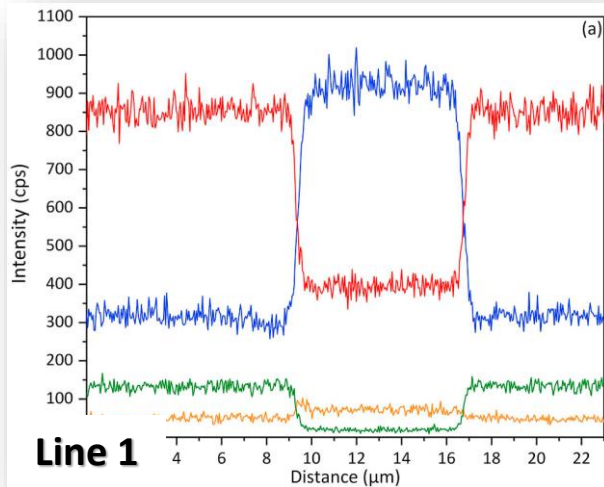
Line 2 – Continuous light-gray phase, Cr- and Mo-rich boride

Cubic, $\text{Cr}_{23}(\text{B}_{1.5}\text{C}_{4.5})$

Line 3 – Discontinuous phase, Mo-rich boride

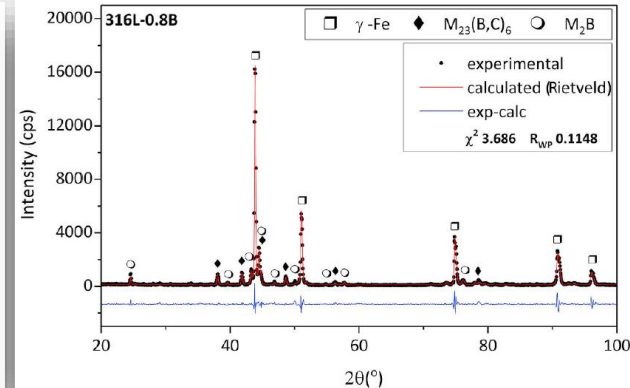
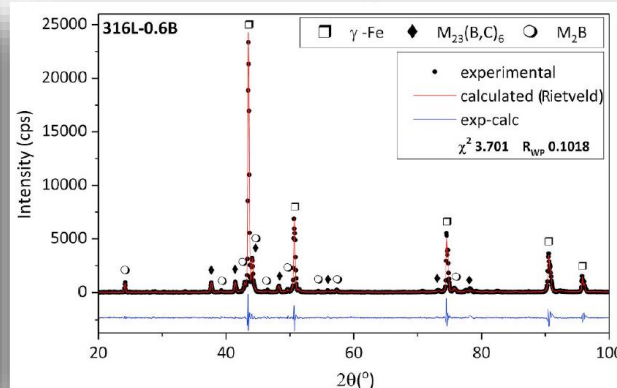
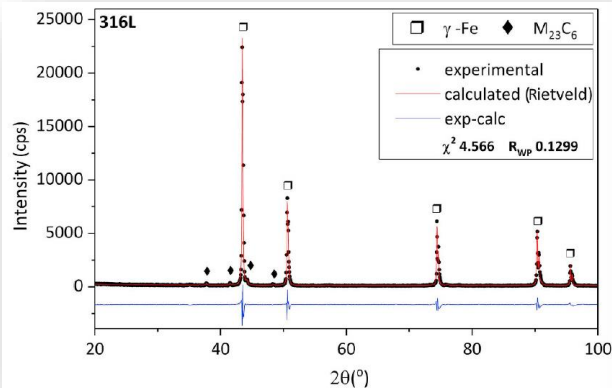
Tetragonal, $\text{Cr}_{1.75}\text{Mo}_{0.25}\text{B}$

— Mo — Cr — Fe — Ni



Crystalline phases

Sample	γ -Fe	$M_{23}C_6$	$M_{23}(B,C)_6$	$Fe_{1.1}Cr_{0.9}B_{0.9}$	$Cr_{1.75}Mo_{0.25}B$
	Cubic	Cubic	Cubic	Orthorhombic	Tetragonal
316L	95.47%	4.53%			
316L-0.6B	78.27%		6.39%	9.40%	5.94%
316L-0.8B	66.21%		15.46%	9.40%	7.91%



Serafini, F.L. et al. (2019) *Materials Characterization*, 152, pp. 253-264.

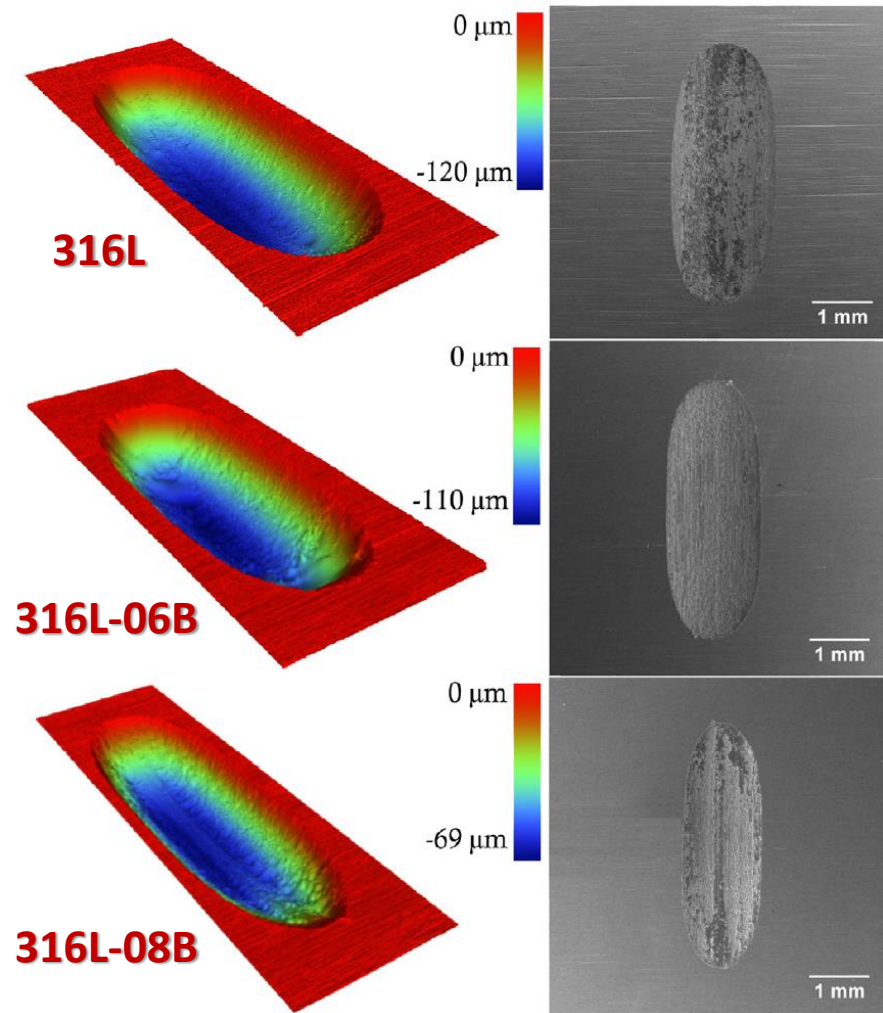
Density and hardness

Sample	Density (g/cm ³)	Vickers Hardness (HV ₁₀)	Instrumented indentation hardness (GPa)		
			Austenite	Dark-grey boride, Cr-rich	Discontinuous boride, Mo-rich
316L	7.13 ± 0.04	89 ± 3	1.6 ± 0.1	-	-
316L-06B	7.37 ± 0.01	159 ± 13	1.9 ± 0.1	20.5 ± 1.3	4.3 ± 0.7
316L-08B	7.35 ± 0.01	174 ± 7	1.9 ± 0.1	22.9 ± 2.1	5.0 ± 0.8

❖ Boron increases the hardness austenitic matrix and creates a network of hard borides along the austenitic grain boundaries

Serafini, F.L. et al. (2019) *Materials Characterization*, 152, pp. 253-264.

Reciprocating sliding wear

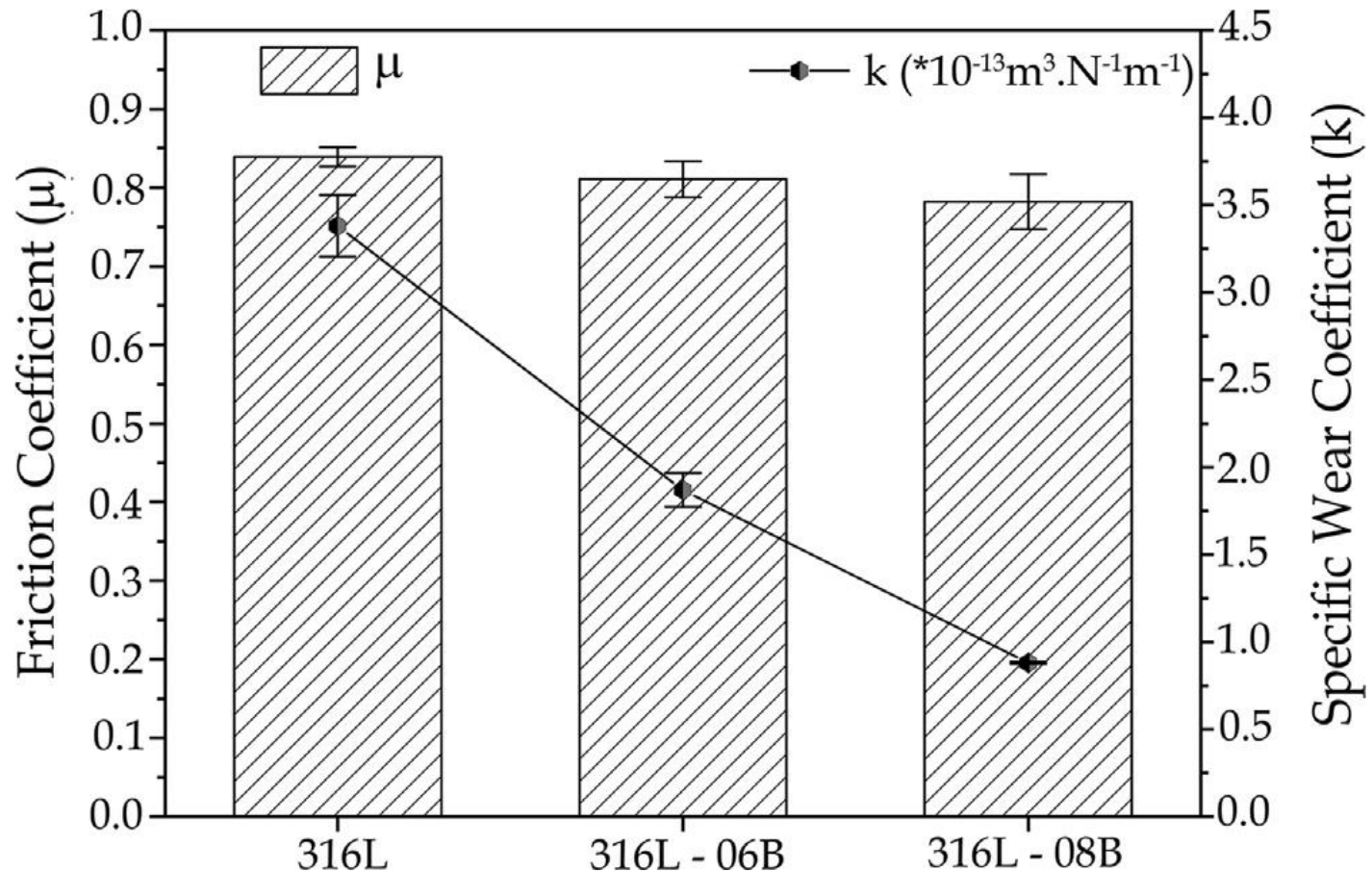


Boron addition

- ❖ Narrower and shallower wear tracks
- ❖ Improved wear resistance (less material removal) that can be related to
 - ❖ Rounded pores (stress concentration regions)
 - ❖ Hard borides (less plastic deformation)

Peruzzo et al. (2019) Wear, 422-423, pp. 108-118.

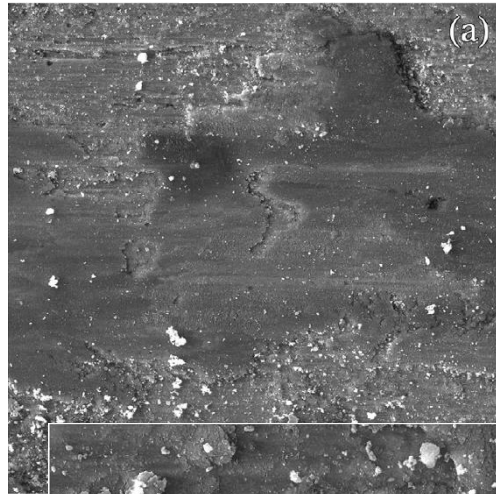
Friction behavior



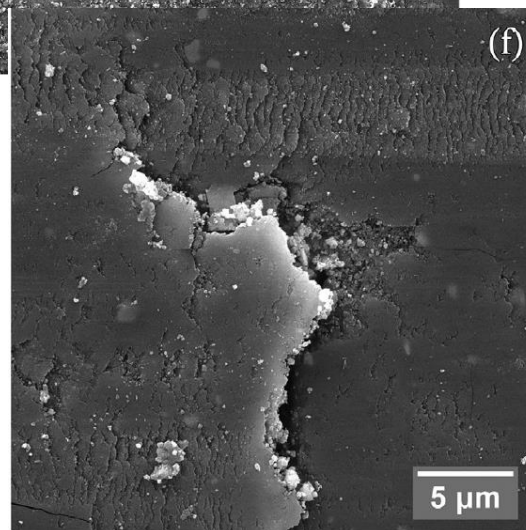
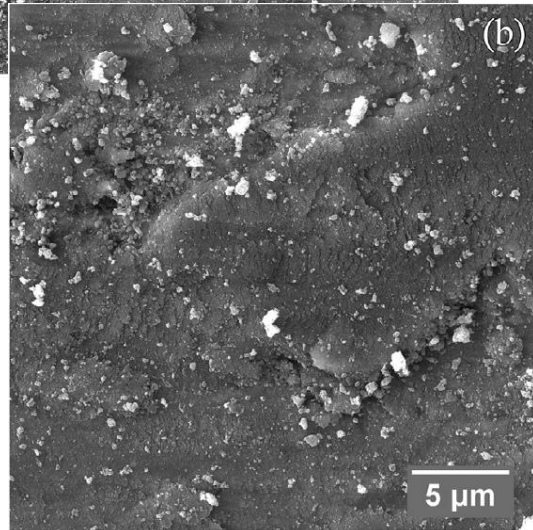
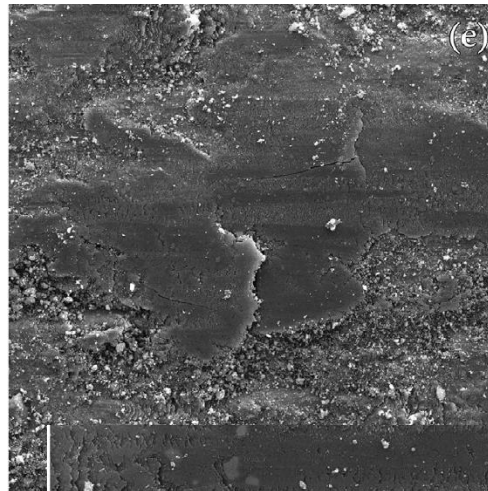
Peruzzo et al. (2019) *Wear*, 422-423, pp. 108-118.

Worn surfaces

316L



316L-08B

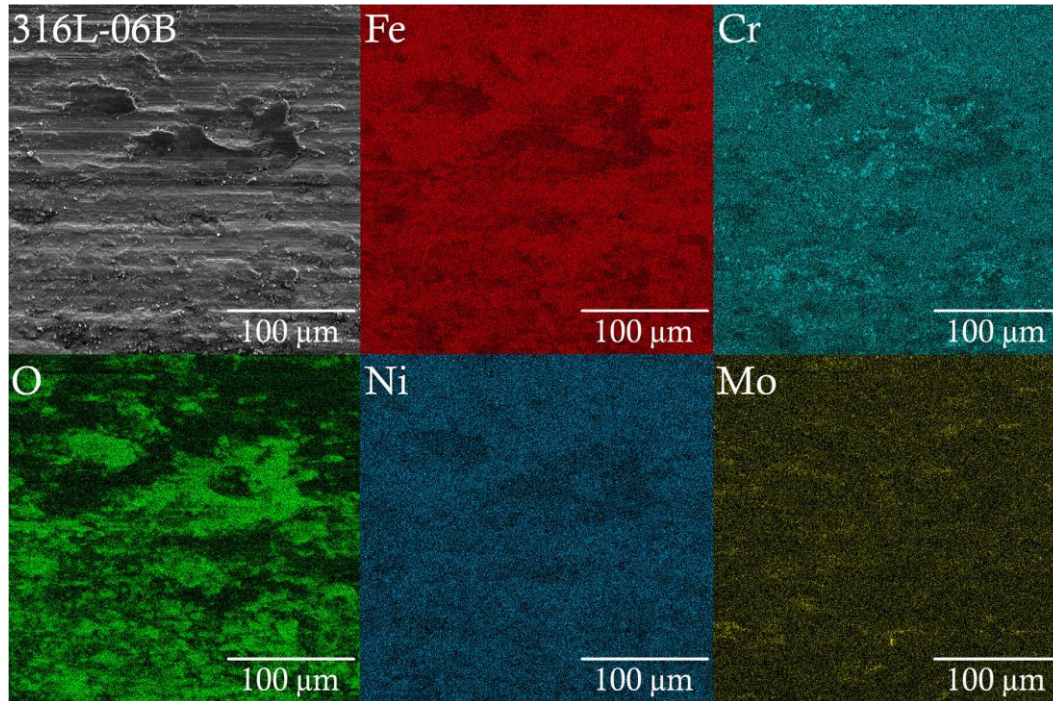


Adhesive wear

plastic deformation,
material transferring and
abrasion grooves

Peruzzo et al. (2019) Wear, 422-423, pp. 108-118.

Worn surfaces



Peruzzo et al. (2019) *Wear*, 422-423, pp. 108-118.

Adhesive wear

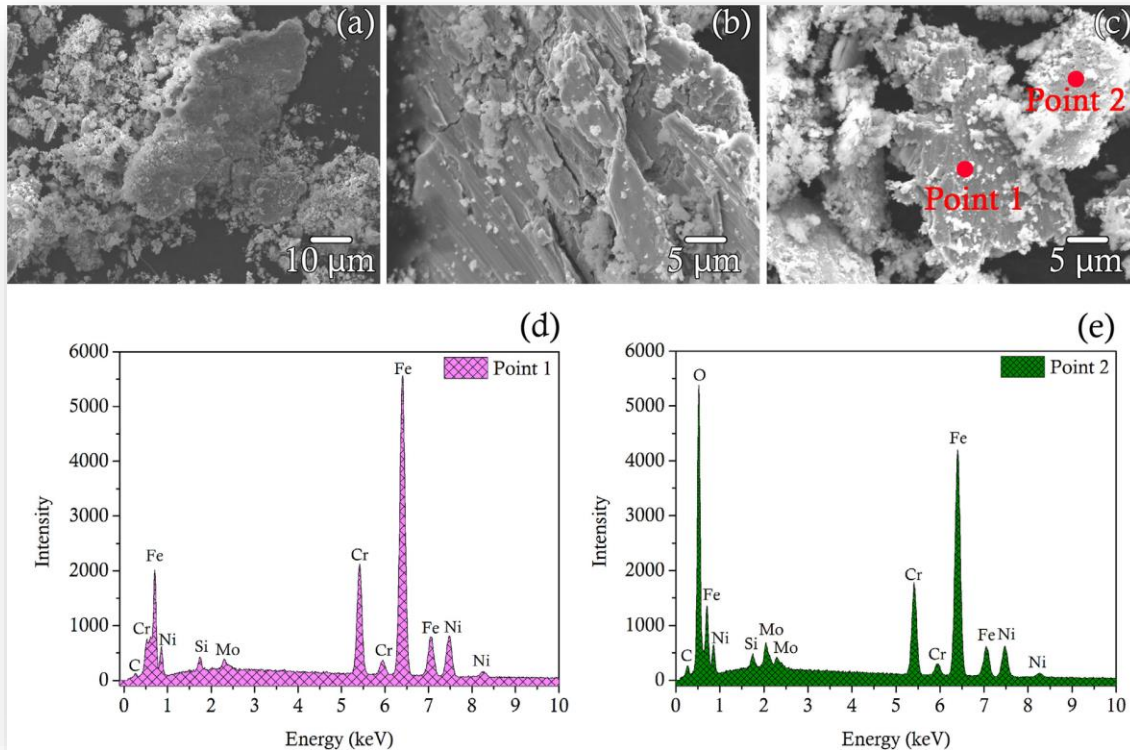
plastic deformation,
material transferring and abrasion
grooves

Oxidative wear by metallic particle oxidation

transfer layer is a mixture of oxide
and metallic particles) and
composed of small clustered
particles

I. Hutchings, P. Shipway, *Tribology: Friction and Wear of Engineering Materials*, 2017.
F.H. Stott, *Tribol. Int.* 31 (1-3) (1998) 61-71.

Wear particles



Wear particles

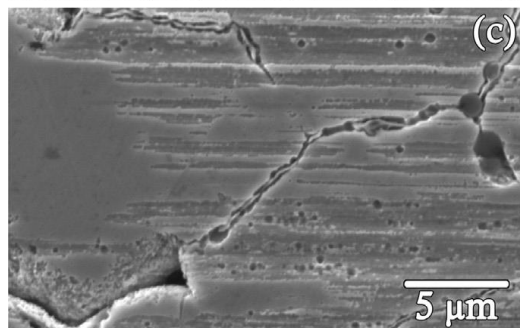
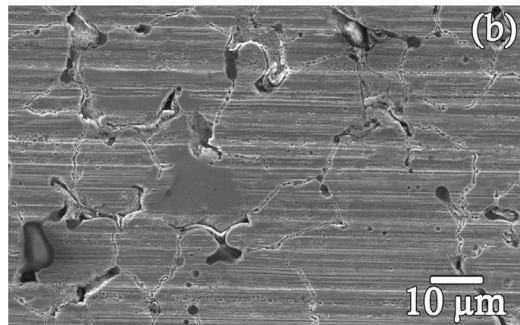
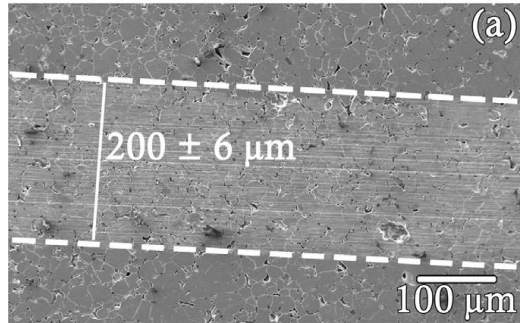
plate and lamellar morphologies

same composition of transfer layer

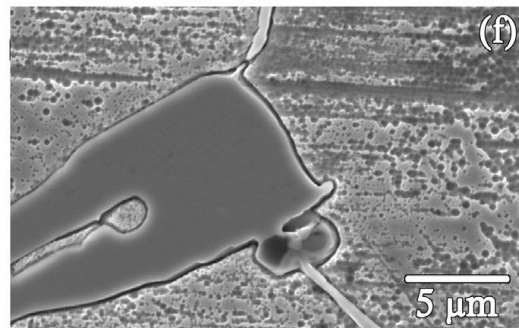
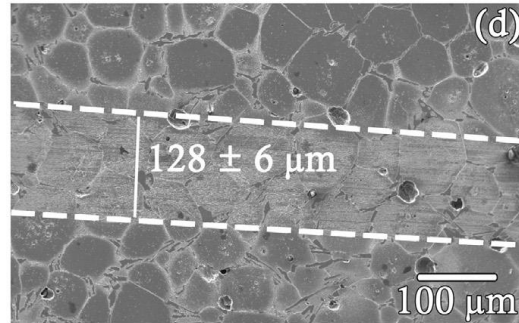
Peruzzo et al. (2019) *Wear*, 422-423, pp. 108-118.

Single-pass sliding test

316L



316L-06B



Single-pass sliding test

Evaluation of load carrying capability of each microstructural constituent (austenite matrix and boron-based precipitates)

Restrict the formation of wear particles, plastic deformation, strain-induced martensitic transformation or oxidation of surface

- ❖ Plastic deformation and abrasion grooves in the austenite matrix
- ❖ Borides are free of any damage

Peruzzo et al. (2019) Wear, 422-423, pp. 108-118.

Single-pass sliding test

Peruzzo et al. (2019) *Wear*, 422-423, pp. 108-118.

Material	E (GPa)	ν^b	H (GPa) ^a	W_E/W_T^a	E^*/H	σ/β^c	ψ	
Austenite matrix	195.30 ± 12.60 ^a	0.30	1.92 ± 0.08	0.15	82.59	2×10^{-4}	1.18	“Plastic contact” $\psi > 0.6$
Cr-rich boride	443.62 ± 18.74 ^a	0.20	20.52 ± 1.27	0.39	13.21	2×10^{-4}	0.18	“Elastic contact” $\psi < 0.6$

Plasticity index

$$\psi = \left(\frac{E^*}{H} \right) \left(\frac{\sigma}{\beta} \right)^{1/2}$$

Material	Dynamic hardness (MPa)	Friction coefficient
316L	510	0.22
316L-06B	1244	0.10

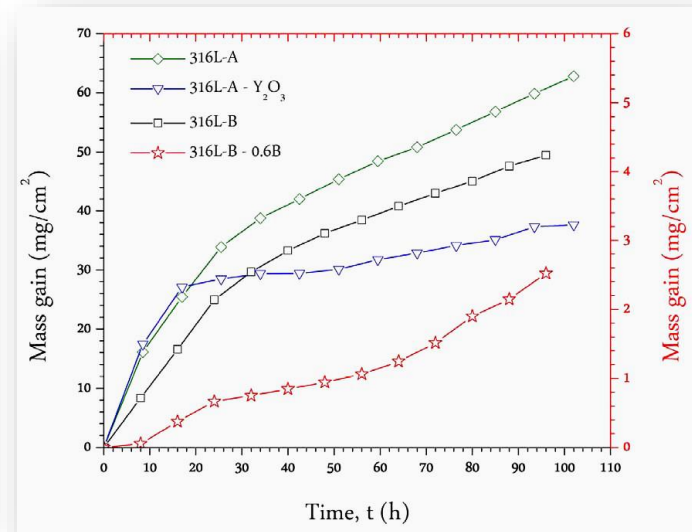
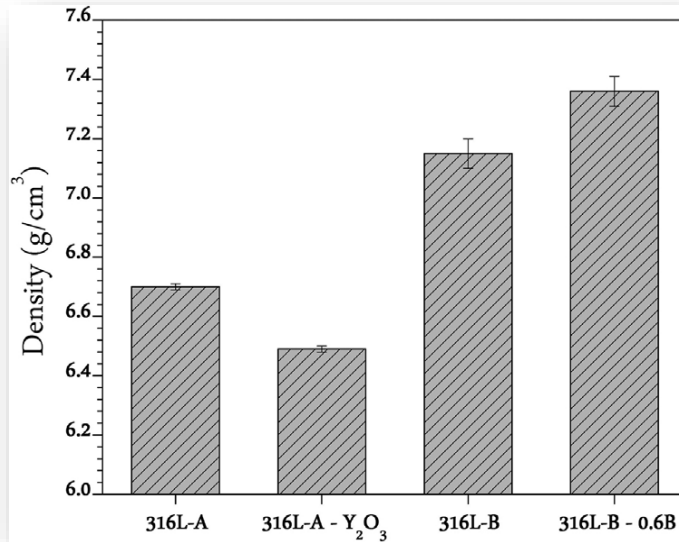
Energy ratio

$$\frac{W_E}{W_T} = \lambda \frac{H}{E^*}$$

Borides have also load carrying capacity

J.A. Greenwood, J.B.P. Williamson, *Proc. R. Soc. A: Math. Phys. Eng. Sci.* 295 (1442) (1966) 300–319.

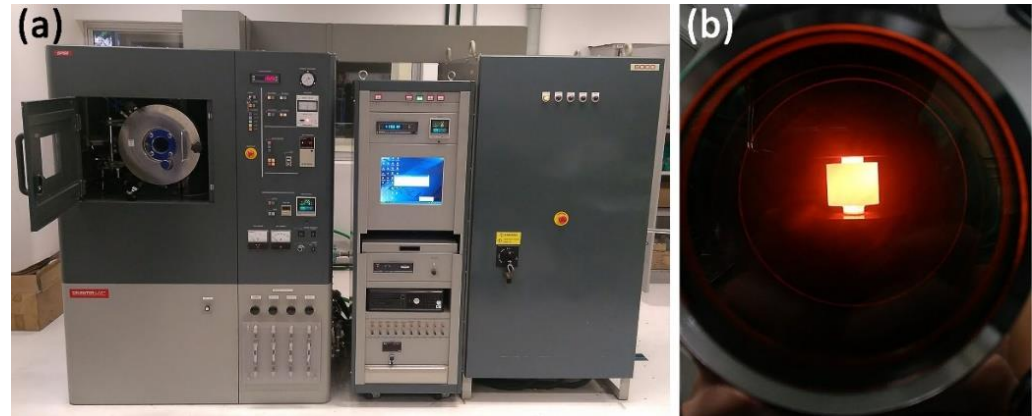
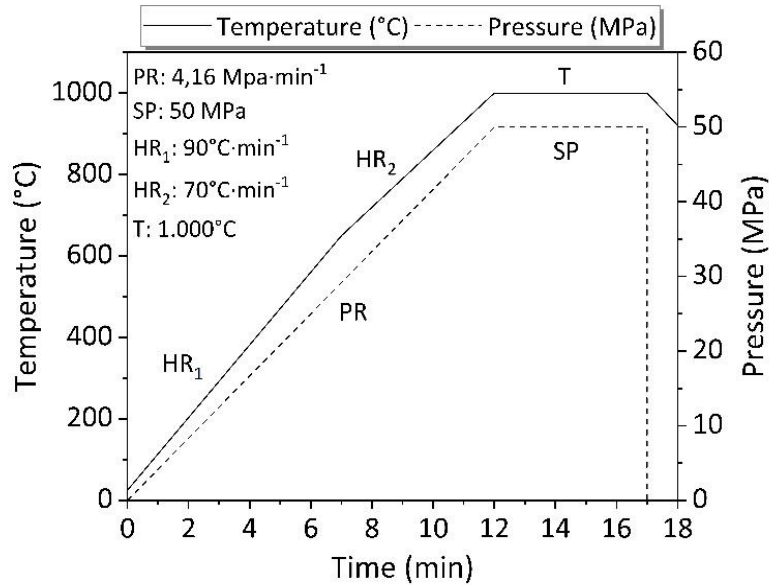
High-temperature cyclic oxidation at 900°C



- ❖ Boron to the 316L steel **strongly improves the cyclic oxidation resistance**
- ❖ **Rounded and smaller pores** contribute with the **reduction** in the **porosity** and the active area for oxidation, (this tendency is more pronounced in the boron-containing steel)
- ❖ **Yttria addition also improves the oxidation resistance** of the steel in a lower ratio than boron

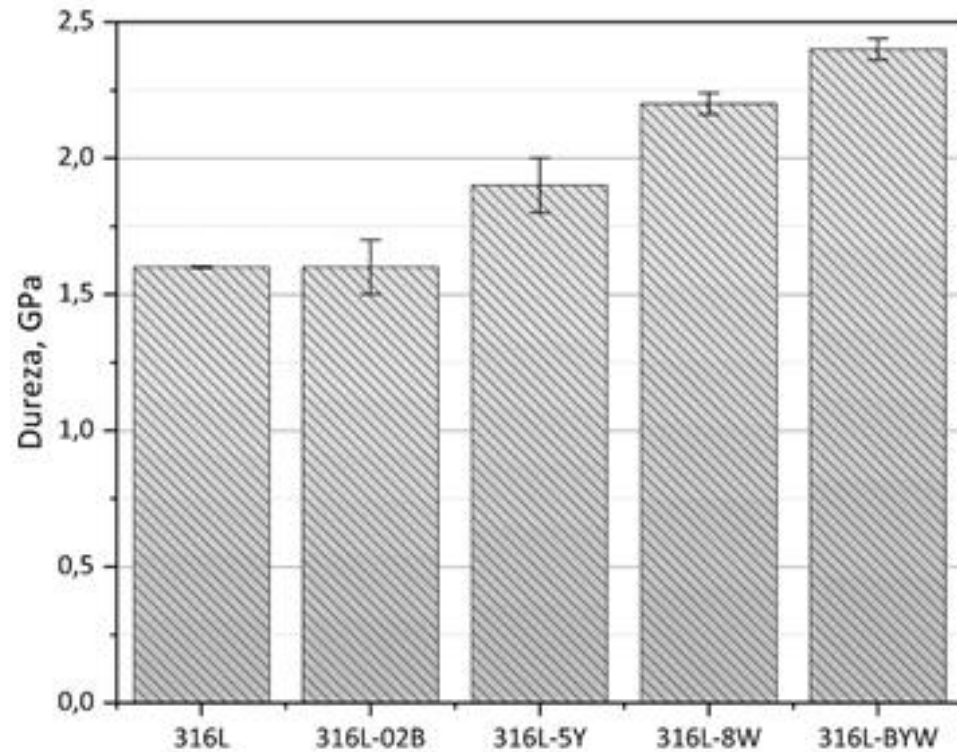
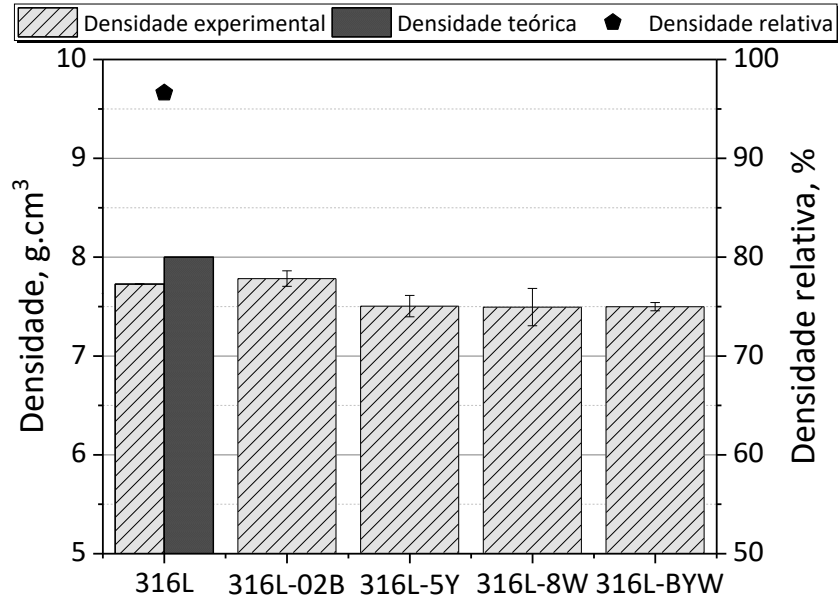
Peruzzo, M., et al. (2017) Corrosion Science, 129, pp. 26-37.

ASS obtained by SPS – recent results



Serafini, F.L. et al. (to be published).

ASS obtained by SPS – recent results



Serafini, F.L. et al. (to be published).

	Base material	Additive type	Additive amount (wt%)	Density (g/cm ³)	Hardness	k mm ³ /Nm	Sintering parameters
[1]	6.90 g/cm ³ 110 HV5	Y ₂ O ₃	1 – 5	6.95 – 6.92	127 – 155	...	1300 °C, 1 h, Hydrogen
[2]	6.90 g/cm ³ 110 HV5	Y ₂ O ₃ Cu Cu-Sn Fe ₃ P Si	3 1 – 3 1 – 3 1 – 2 1 – 5	...	122 104 – 100 135 – 127 171 – 218 85 – 80	...	1300 °C, 1 h, Hydrogen
[3]	6.90 g/cm ³ 38 HRB 6.1x 10 ⁻¹³ mm ³ /Nm	Y ₂ O ₃ -B ₂ Cr Y ₂ O ₃ -BN	5-2 5-1	7.35 7.05	74 63	3.5x 10 ⁻¹³ 4.7x 10 ⁻¹³	1250 °C, 30 min, Vacuum
[4]	6.5 g/cm ³	Y ₂ O ₃	10	6.18	1250 °C, 60 min, Hydrogen
[5]	6.61 g/cm ³ 82 HV5	Y ₂ O ₃	3 – 8	6.35 – 6.08	118 – 125	...	1350 °C, 1 h, Hydrogen

[1] S. Lal, G.S. Upadhyaya. J. Mater. Sci. 24 (9) (1989) 3069–3075.

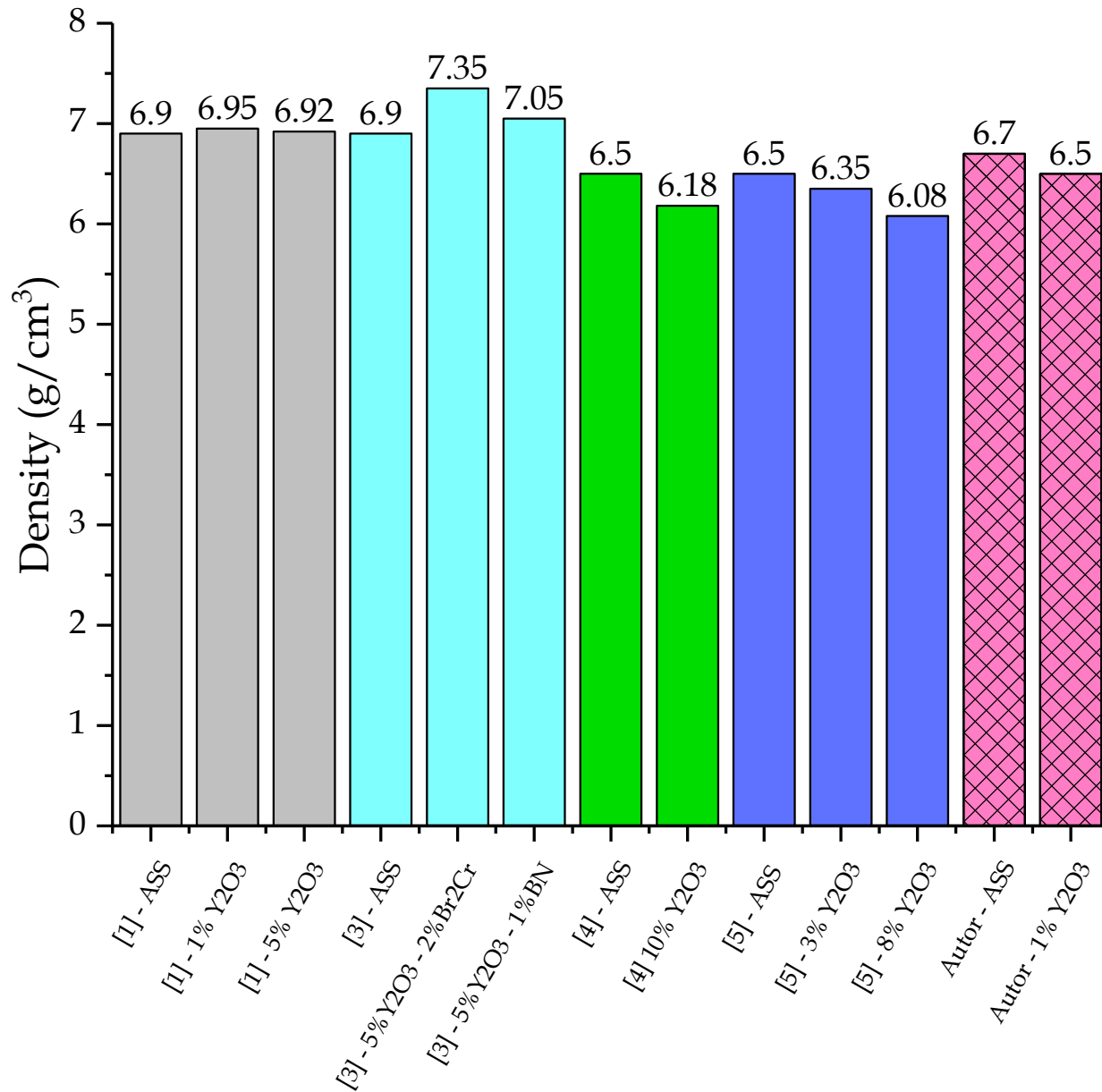
[2] S. Lal, G.S. Upadhyaya. Sol. Sta. Phen. 8-9 (1990) 361–368.

[3] M. Vardavoulias, et al. Tribol. Int. 29 (6) (1996) 499–506.

[4] J. Shankar, et al. Corros. Sci. 46 (2004) 487–498.

[5] A. Raja Annamalai et al. Corros. Eng. Sci. Technol. 50 (2015) 91–102.

Data compilation
Properties of P/M ASS



Data compilation

Effect of additives in density of P/M ASS

Sintered Austenitic Stainless Steel

Final remarks

- Boron can be used either **elemental** or in **compound** (FeB , Fe_2B , NiB and Cr_2B) as an additive in iron-based systems and stainless steels
- Boron acts as a **sintering enhancer**, and sintering temperature can be reduced to about $1240\text{ }^\circ\text{C}$
- The addition of boron in elemental form results in the **good properties** and structure for stainless steel
- Boron addition up to **0.8 wt%** for 316L stainless steel powders **increases density, mechanical properties, corrosion, and wear resistance**. Nearly full densification was obtained with enough eutectic phase formation
- Yttria addition improves high-temperature oxidation of P/M ASS.
- *The resultant parts might be used in applications where wear and high-temperature resistance are desired*

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